Sulfide toxicity



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What it does

Sulfide toxicity reduces nutrient uptake of plants by reducing root respiration. It has an adverse effect on metabolism, particularly when an excessive amount is taken up by the rice plant.

Why and where it occurs

Sulfide toxicity is not very common in rice. It is, however, associated with low-Fe soils.

It can occur in well-drained sandy soils, degraded paddy soils, poorly drained organic soils, and acid sulfate soils.



Sulfide toxicity causes roots to be coarse and sparse

How to identify

Check the field for the following symptoms:

- Interveinal chlorosis of emerging leaves coarse, sparse, dark brown to black root system.
- Freshly uprooted rice hills often have poorly developed root systems with many black roots (stains of Fe sulfide) unlike healthy roots, which are covered with a uniform and smooth orange-brown coating of Fe³⁺ oxides and hydroxides.
- Sulfide toxicity can also cause increased occurrence of diseases, such as brown spot. Leaf symptoms of sulfide toxicity are similar to those of chlorosis caused by Iron (Fe) deficiency. Other diagnostic criteria are similar to those of Fe toxicity, but has different visual leaf symptoms.

How to manage

- Sulfide tolerant plant varieties.
- Avoid continuous flooding and use intermittent irrigation in soils that contain large concentrations of S, have high organic matter status, and are poorly drained
- Carry out dry tillage with proper machines after the rice harvest to enhance Fe oxidation during the fallow period.
- Balanced use of nutrients (NPK or NPK+ lime) to avoid nutrient stress and improve root oxidation power
- Apply sufficient K fertilizer
- Avoid using excessive amounts of organic residues (manure, straw) in soils containing large amounts of Fe and organic matter, and in poorly drained soils

Where possible,

In temperate climates, coat seeds with oxidants (e.g., Calcium (Ca) peroxide) to increase the 0_2 supply and improve seed germination

 Apply K, P, and Mg fertilizers, apply Fe (salts, oxides) on low-Fe soils to increase immobilization of H₂S as FeS







